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A neutralization reaction occurs between an acid and a base and produces water and a salt. The final solution is closer to a neutral pH than either the acid or base alone. Hydrochloric acid (HCl) reacts with potassium hydroxide (KOH), producing a solution of potassium chloride (KCl) and water (H₂O).

[Answers Acids, Bases, and Solutions](#)

“ Acids, Bases and Salts ” focuses on acids and bases, and Salts in solutions. These NCERT Solutions comprises answers to all the questions of the chapter that are there in the NCERT textbook. We...

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7 Acids, bases and ions in aqueous solution . 8 Reduction and oxidation . 9 Non-aqueous media . 10 Hydrogen . 11 Group 1: the alkali metals . 12 The group 2 metals . 13 The group 13 elements . 14 The group 14 elements . 15 The group 15 elements . 16 The group 16 elements . 17 The group 17 elements . 18 The group 18 elements

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Hard and Soft Acids and Bases (HSAB) Principle is a qualitative concept introduced by Ralph Pearson to explain the stability of metal complexes and the mechanisms of their reactions. However, it is also possible to quantify this concept based on Klopman's FMO analysis using interactions between HOMO and LUMO.

[HSAB PRINCIPLE - PEARSON'S HARD & SOFT ACIDS & BASES ...](#)

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