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Dimensional Close Packing Trick To Calculate
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solid state part 5 Ch 1 The Solid State +
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~~SOLUTION (intext questions 1.1 to 1.24)~~

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Atoms in a Unit Cell 1.6 Close Packed [...]

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????? ???? ???? ????? ?? ???? ??? ??????

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www.ncerthelp.com (Visit for all ncert solutions in text and videos, CBSE syllabus, note and many more) Chemistry Notes for class 12 Chapter 1 The Solid State Solids Solids are the chemical substances which are characterised by define shape and volume, rigidity, high density, low compressibility. The constituent particles (atoms, molecules or ions)

Chemistry Notes for class 12 Chapter 1 The Solid State

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This chapter deals with the solid state. It is comprised of 11 subtopics. This chapter deals with crystalline and amorphous solids, as well as imperfections in solids. It explains the unit cell in detail, and then builds up to explain the solid state. The solutions are made in accordance with the ...

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(Updated for 2020-21)*

Solid state chemistry class 12 ncert solutions At sufficiently low temperatures the thermal energy is low and the intermolecular forces close them so much that they attach to each other and the definite positions are absorbed. It can still oscillate around its mean positions and the substance remains in solid-state.

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12 Chemistry. Solid State NCERT InText Solution. Question: 1.1 - Why are solids rigid? Answer: The particles of solids are close packed and can only oscillate about their fixed positions. These properties make

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solids rigid. Question: 1.2 - Why do solids have a definite volume?

12 Chemistry Solid State NCERT In Text

Solution part 1

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NCERT Class 12 Chemistry Chapter 1 The Solid State Notes ...

Chemistry 2 In Class XI you have learnt that matter can exist in three states namely, solid, liquid and gas. Under a given set of conditions of temperature and pressure, which of these would be the most stable state of a given substance depends upon the net effect of two opposing factors.

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