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A change in physical form of a substance or
physical properties of a substance like color,
melting point, state, size, shape, and
arrangement of molecules is known as

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physical change. A change in chemical properties of a substance like composition, reactivity with water, corrosiveness, and pH is known as chemical change. In chemical change, a new substance will be formed.

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Change • Chapter 9 145 CHAPTER 9
SOLUTIONS MANUAL 35. Aqueous
solutions of potassium iodide and silver
nitrate are mixed, forming the precipitate
silver iodide. chemical equation: $KI(aq)$
 $AgNO_3(aq) \rightarrow KNO_3(aq) AgI(s)$ complete
ionic equation: $K^+(aq) I^-(aq) Ag^+(aq) NO_3^-(aq)$
 $\rightarrow K^+(aq) NO_3^-(aq) AgI(s)$

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12
Chemical Reactions

Solution. Physical: boiling and melting are physical changes. When water boils, no bonds are broken or formed. The change could be written: $\text{H}_2\text{O} (l) \rightarrow \text{H}_2\text{O} (g)$

Chemical: The dark grey nail changes color

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to form an orange flaky substance (the rust);
this must be a chemical change.

3.6: Changes in Matter: Physical and
Chemical Changes ...

240 Chemistry: Matter and Change •

Chapter 12 Solutions Manual CHAPTER 12
SOLUTIONS MANUAL 30. Compare and

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contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

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