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$$\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$$

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amount of blue

copper sulphate in a
clean test tube, then

heat the content of
the test by using a

flame, A black

substance will be

formed, Blue copper

sulphate decomposes

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colour) and sulphur
trioxide. $\text{CuSO}_4 \cdot \text{CuO}$
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